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Nuclear and electron spin relaxation in paramagnetic systems

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Paramagnetic materials

- Paramagnetic materials have positive magnetic susceptibility, associated with unpaired electrons
- Paramagnetic solutions contain free radicals or transition metal ions/complexes. Oxygen gas (triplet ground state) is also paramagnetic
- Unpaired electron has large magnetic moment, about 650 times that of proton
- This large magnetic moment affects strongly NMR properties, not least relaxation
- · Electron spin is strongly coupled to lattice













- Important systems: transition metal & lanthanide ions and complexes
- ESR relaxation often dominated by ZFS
- If the metal ion in low-symmetry complex (lower than O_h or T_d), static ZFS, can be modulated by rotation
- Hydrated metal ions: *transient* ZFS modulated by collisions (distortions of the solvation shell)



Generalized BM

- For S ≥ 3/2, the electron spin relaxation is expected to be multiexponential – can be handled within Redfield limit
- Systems with static and transient ZFS can be handled within Redfield limit, $\Delta_t^2 \tau_v^2 \ll 1$, $\Delta_s^2 \tau_R^2 \ll 1$





- Macroscopic/microscopic
- Complications by exchange
- P_M : mole fraction bound
- τ_M : exchange lifetime
- Subscript *M*: in-complex properties
- Subscript *P*: measured properties
- T₁ most common
- Fast exchange for T_1 : $\tau_M << T_{1M}$















- SLE-based, calculations in frequency domain (J(ω))
- Nuclear spin interacts with a "composite lattice", containing electron spin. The lattice described in terms of electron Zeeman, transient & static ZFS, reorientation & distortion (pseudorotation)
- Calculation of PRE involves setting up and inverting a very large matrix representing the lattice Liouvillean in a complicated basis set. Computationally heavy
- Very general, can be used as benchmark for simpler models
- Equivalent to the Grenoble model (formulated in time domain, G(t))







Susceptibility anisotropy

- If susceptibility is anisotropic, the magnitude of the electron spin magnetic moment along the field direction orientation-dependent
- The local dipolar field around the average magnetic moment is then *not averaged to zero* by rotation
- The non-zero averaged dipolar field leads to extra shifts of nuclear spin resonances, which depend on position of the nucleus in molecular frame *pseudocontact shifts*









- System: propylene glycol + 4-oxo-TEMPO-d₁₆
- Two isotope species: ¹⁴N (*I*=1) &¹⁵N (*I*=1/2)
- ESR lineshapes and NMRD (10 kHz-20 MHz)
- ESR interpreted using SLE & Redfield
- NMRD interpreted with a recent theory (Kruk *et al.* JCP 2013), Redfield limit







- ESR lines split by isotropic hyperfine interaction
- Electron spin relaxation caused by anisotropic hyperfine (dipolar) with *nitrogen* spin & Δg, modulated by rotation
- Solvent proton relaxation caused by outer-sphere dipolar interaction with electron spin
- Modulation by translational diffusion (dominates at high temp) and electron relaxation (important @ lower temp)

















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Application to paramagnetic proteins

- Paramagnetic proteins: one measures on protein surrounding the metal center, usually without exchange
- More demanding, very severe broadening of NMR signals in the vicinity of the metal
- *T*₂⁻¹ proportional to γ₁², protons much more broadened than ¹³C or ¹⁵N, "protonless NMR"

Copper-traficking protein

- ...from Pseudomonas Syringae, binds Cu(II)
- Apo-protein and Cu(I) (diamagnetic) analogue, structures known
- Study of electron spin relaxation through NMRD of (exchanging) water protons
- No proton signals closer than about 11 Å from Cu(II)
- ¹³C PRE, PCSs, along with ¹H NOEs
- · Paramagnetic constraints necessary to locate the copper





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